

A review on chemical structure of natural dye pigment

¹Prajapati Mayur ²Modi Nainesh

¹Research Scholar, ²Associate Professor

^{1,2}Department of Botany, Bioinformatics and Climate Change Impacts Management,

School of sciences, Gujarat University, Navrangpura, Ahmedabad-380009

E-mail – ¹mayur.prajapati1807@gmail.com, ²nrmodi@gujaratuniversity.ac.in

Abstract: Natural dyes are used for coloring of food materials, leather and fibers like wool, silk and cotton. Recently, many commercial dyers have started using natural dyes to overcome the environmental mutilation caused by artificial dyes. Pigments are present in all living matter which offering attractive colours. The pigments can be classified as natural, synthetic or inorganic by their origin. Natural pigments also classified based on their chemical structure such as indigoid, Anthraquinone, Alpha-hydroxy-naphthoquinones, Flavonoids, Dihydropyrans, Anthocyanidins and Carotenoids. This paper reports the studies available on chemical structure of natural dye pigments from different plant species

Key Words: Natural dye, Pigments, Chemical structure.

1. INTRODUCTION:

A dye is a coloured substance that chemically bonds to the substrate to which it is being applied. Natural dyes are used for coloring of food materials, leather and fibers like wool, silk and cotton. The use of environmental friendly natural dyes on textiles have become a matter of significant importance due to the increased environmental consciousness in order to avoid some harmful synthetic dyes [Samantha *et al.*2009]. These dyes are extracted from vegetable and animal matter with no or very minute chemical processing. Recently, many commercial dyers have started using natural dyes to overcome the environmental mutilation caused by artificial dyes [Arora *et al.*2017]. Natural colorants are also derived from naturally occurring sources like insects (e.g., cochineal beetles and lac scale insects); animals (e.g., some species of mollusks or shellfish); and minerals (e.g., ferrous sulfate, ochre, and clay) without any chemical treatment [Chengaiyah *et al.*2010]. Germany was the first to take initiative to put ban on numerous specific azo dyes for their manufacturing and applications. The Netherlands, India and some other countries have also followed with a ban on azo dyes [Patel *et al.*,2011]. Overexploitation of natural resources to obtain dyes may result in deforestation and threaten endangered species. For these reasons, the Global Organic Textiles Standard (GOTS) permits the use of safe synthetic dyes and prohibits the use of natural dyes from endangered species [Saxena *et al.*,2014]. Natural colouring agents are phytochemicals which are mainly obtained from natural plant resources and they are very beneficial to human health. [Leong, H. Y *et al.*2018].

1.1. Plant Pigments:

A pigment is a substance which proficient of absorbing light, so it could be considered pigments almost all substances [Hoffman *et al.*,2009]. Usually, pigment absorbs very specific wavelengths, and emit a characteristic electromagnetic signal that allows us to identify them. Most often chemical compounds selectively absorb within certain radiation ranges and the color of a given compound is complementary to the absorbed radiation [Herrera.,2015]. Pigments are present in all living matter which offering attractive colours and playing fundamental roles in organism's growth [Delgado-Vargas *et al.*,2000].

1.2. Classification of Pigments:

1.2.1. Based on their origin:

The pigments can be classified as natural and synthetic by their origin. Natural pigments are produced by living organisms such as plants, animals, fungi and microorganisms. Synthetic pigments are developed from laboratories. The organic compounds are both natural and synthetic pigments. Inorganic pigments can be found in nature or synthesized [Bauernfeind *et al.*,1981].

1.2.2. Based on the Chemical Structure [Vankar *et al.*,2000]:

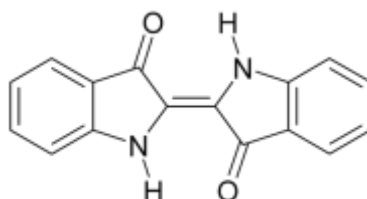


Figure 1. Indigo [Vankar *et al.*,2000]

Indigoid dyes: This is the most important group of natural colours, derived from *Indigofera tinctoria* [Vankar et al.,2000].

1.2.3. Anthraquinone dye:

Some of the major red dyes are based on the structure of anthraquinone. These are derived from both plants and insects. Such colors are distinguished by good light fastness. They form complexes with metal salts, and the resulting metal-complex coloring has good wash quality [Vankar et al.,2000]

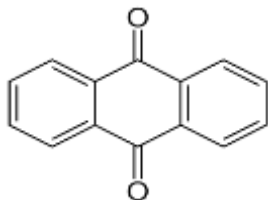


Figure 2. Anthraquinone [Vankar et al.,2000].

1.2.4. Alpha-hydroxy-naphthoquinones:

The most prominent member of this type of coloring is lawsone or henna from *Lawsonia inermis* [Vankar et al.,2000].

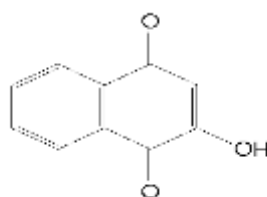


Figure 3. Alpha-hydroxy-naphthoquinone [Vankar et al.,2000].

1.2.4. Flavonoids:

Flavonoids are a diverse group of polyphenolic compound contribute to the yellow color of horticultural products. These are widely distributed in the plant kingdom and more than 4000 structurally distinct flavonoids in plant sources have been identified [Patel et al.,2008]. Flavonoids can be subdivided into different subgroups depending on the C ring carbon attached to the ring and the degree of insaturation and oxidation of the C ring. Flavonoids are called isoflavones, in which the B ring is connected in position 3 of the C ring. Those in which the B ring is connected in position of 4 are considered neoflavonoids, while those in which the B ring is connected in position 2 are further subdivided into several subgroups based on the structural features of the C ring. Such subgroups include: flavones, flavonols, flavanones, flavanonols, catechins or flavanols, anthocyanins and chalcones [Panche et al.,2016].

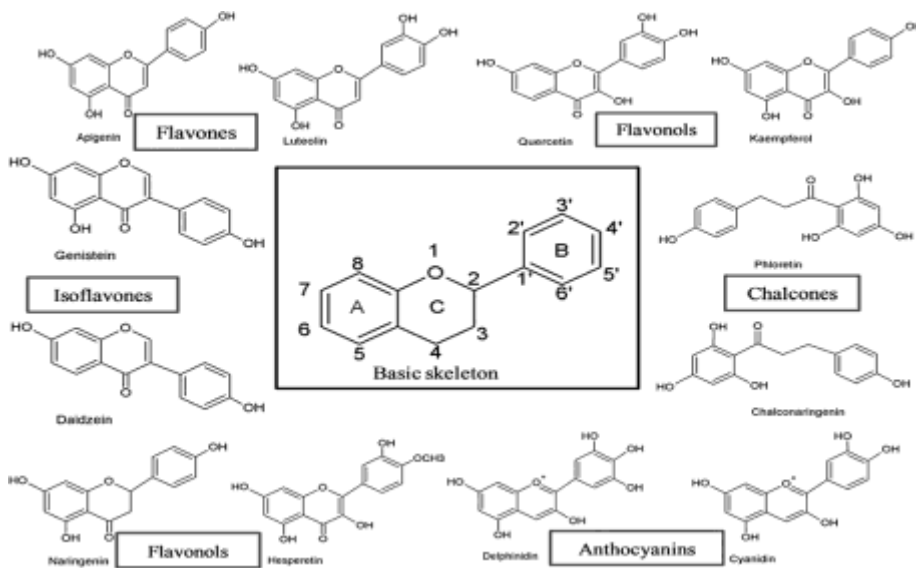


Figure 4. Types of flavonoids [Panche et al.,2016]

1.2.5. Dihydropyrans:

Di-hydropyrans, viz. haematin and its leuco form, haematoxylin are closely related to the flavones in the chemical structure. These are important natural dyes on silk, wool and cotton for the dark shades. The common examples are logwood, Brazil wood and sappan wood [Srivastava et al.,2019].

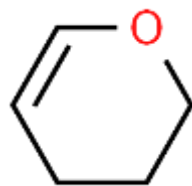


Figure 5. Dihydropyrans

1.2.6. Anthocyanidins:

Anthocyanins (in Greek, antho means herb, kyanos means blue) are glycosides of 2 (in all other forms, flavilium ionic form is stable), strongly colored sap responsible for various colors in higher plants. These are probably the best known natural food colorants as they are widely distributed in the plant kingdom and range in color from apple red to grape blue to brinjal violet [Prabhu *et al.*,2012]. Because of its physicochemical properties (eg poor stability and low solubility), anthocyanidins (aglycone) are most commonly found in nature as their glycosidic forms of salts, known as anthocyanins [Celli *et al.*,2019]. In addition to their structure, pH, temperature [Sui *et al.*,2014], oxygen [Weber *et al.*,2017], light [Mahdavi *et al.*,2016], water interaction [Tonon *et al.*,2010] and other compounds (e.g. ascorbic acid) also influence the stability of anthocyanins [Guldiken *et al.*, 2017].

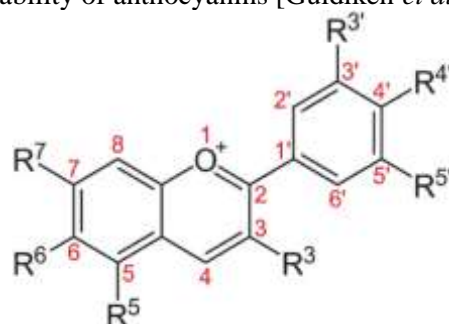


Figure 6. Basic Anthocyanins structure [Guldiken et al.,2017]

1.2.7. Carotenoids:

Carotenoids that are widely distributed in plants and animals are expressed by linearly conjugated polyene tetraterpenes. The carotenoids are brightly coloured pigments in which the highly conjugated π -electron system confers different colours like yellow, orange and red to the molecule. The name carotene is derived from the orange pigment found in carrots (separated in 1831, Wackenroder) [Wackenroder 1831]. In nature, carotene (e.g. β -carotene from *Daucus carota*) and xanthophyll (e.g. Lutein from *Tagetes erecta*) are two forms of carotenoids [Piccaglia et al.,1998].

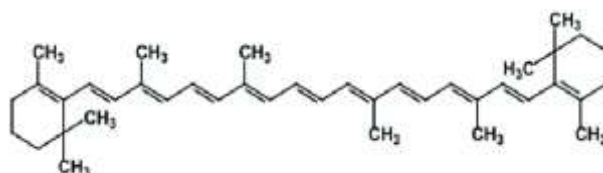
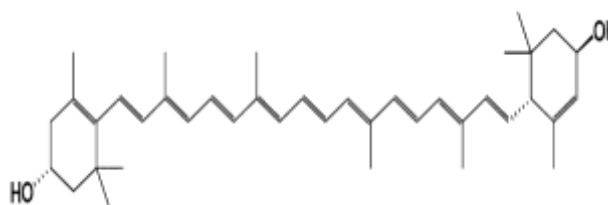
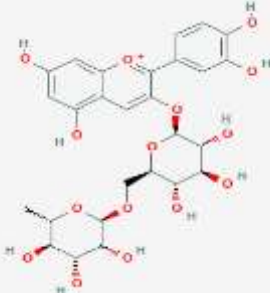
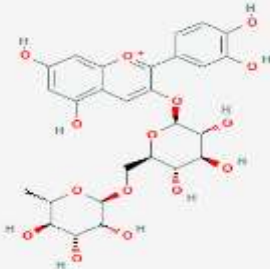
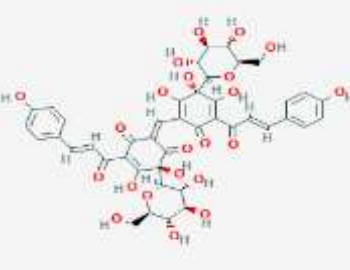
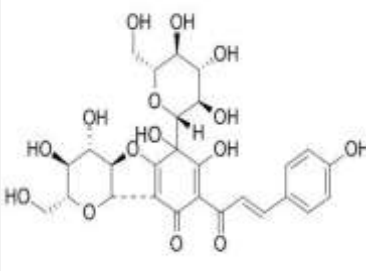
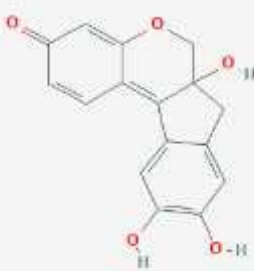
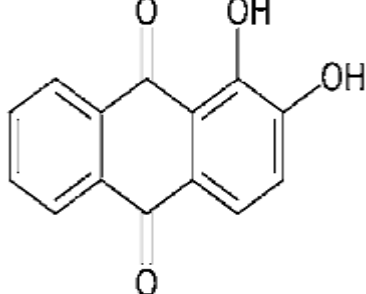
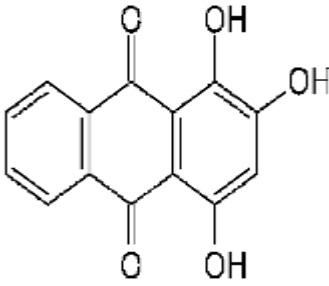
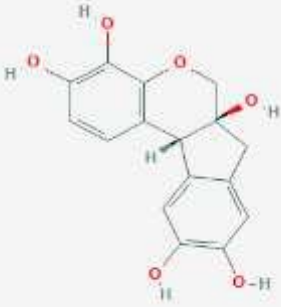
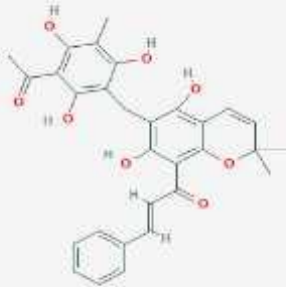
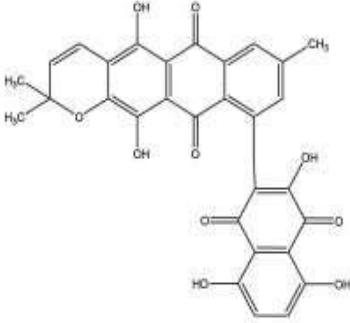
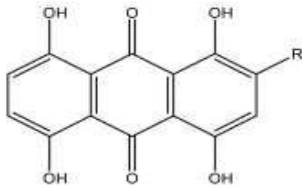
Figure 7. Structure of β -carotene

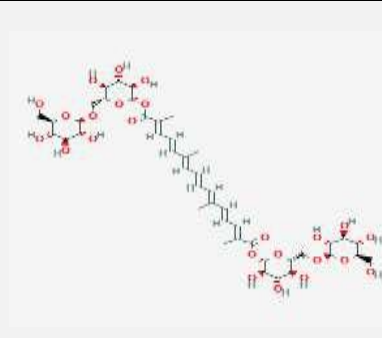
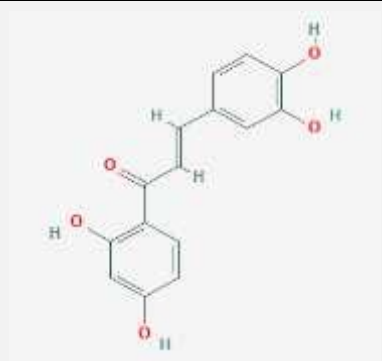
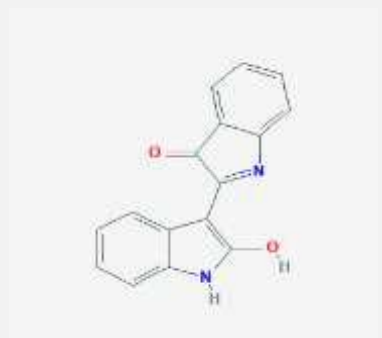

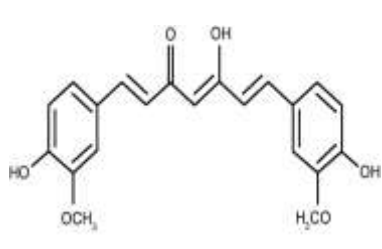
Figure 8. Structure of Lutein

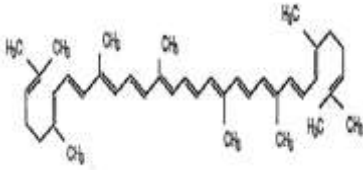
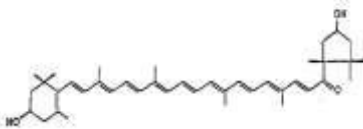
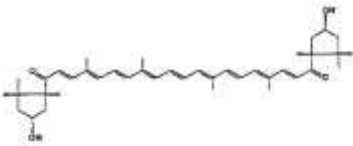
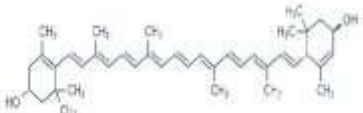
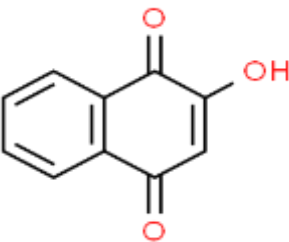
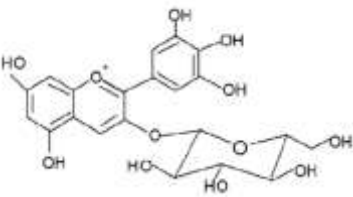
Table 1: Some Important dye-yielding plants with pigments and their chemical properties

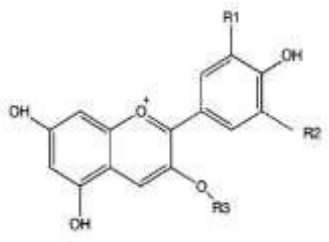
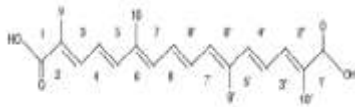

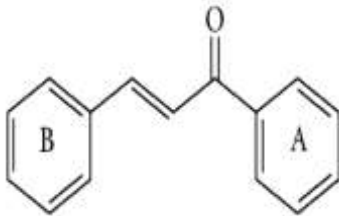
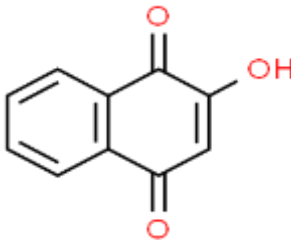
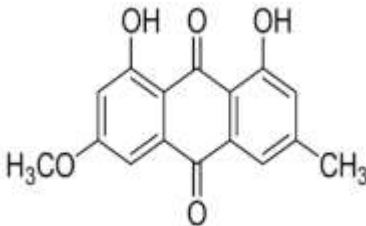
Name of Plants	Part of plant	Name of pigments	Colour of pigment	Chemical structure	References

<i>Brachychiton acerifolias</i>	Flower	1. Cyaniding 3-O-rutinoside		(24), (25)
		2. Cyanidin 3-O-glucoside		(24), (26)
<i>Carthamus tinctorius</i>	Petal	1. Carthamin		(27), (28)
		2. Safflor yellow		(27), (29)
<i>Caesalpinia sappan</i>	Wood	Brazilein		(30), (31)
<i>Rubia tinctorium</i>	Root	1. Alizarin		(32), (33)

		2. Purpurin		(32), (33)
<i>Haematoxylon campechianum</i>	wood	Haematoxylin		(34), (35)
<i>Mallotus philippinensis</i>	Fruit	Rottlerin		(36), (37)
<i>Tectona grandis</i>	Leaf	Tectograndone		(38), (39)
		Tectoleafquinone	 $R = \text{H}_2\text{C}=\text{CH}-\overset{\text{CH}_3}{\text{C}}=\text{CH}_2$	(39), (40)

<i>Corcus sativus</i>	Flower (Stigma)	Crocin		(41), (42)
<i>Butea monosperma</i>	Petal	Butein		(43), (44)
<i>Indigofera tinctoria</i>	Leaf	Indirubin		(45), (46)
<i>Srrobilanthes flaccidifolius</i>	Leaf	Indigo		(47), (48)
<i>Curcuma longa</i>	Rhizome	Curcumin		(49), (50)

<i>Solanum lycopersicum</i>	Fruit	Lycopene		(51), (52)
<i>Capsicum annuum</i>	Fruit	1. Capsanthin		(49), (53)
		2. Capsorubin		(49), (53)
<i>Tagetes erecta</i>	Flower	Lutein		(54), (55)
<i>Lawsonia inermis</i>	Leaf	Lawsone		(56), (57)
<i>Garcinia indica</i>	Fruit	1. Cyanidin 3-glucoside		(58), (59)

		2. Cyanidin 3-sambubioside		(58), (60)
<i>Nyctanthes arbortristis</i>	Flower (Calyx)	Crocein		(61)
<i>Basella rubra</i>	Fruit	Gomphrenin I		(62), (63)
<i>Bauhinia purpurea</i>	Flower (Petal)	Chalcone		(64), (65)
<i>Woodfordia fruticosa</i>	Leaves	Lawsone		(57), (66)
<i>Rheum rhabarbarum</i>	Root	Physcion		(67), (68)

2. CONCLUSION:

Natural dyes not only have the ability of dyeing but also have a wide range of medicinal properties. We can extract different types of natural pigments from the plants. Such natural pigments will show advantages over synthetic

pigments. Mostly these natural pigments are used for dyeing of different fabrics. Natural pigments are less harmful, less hazardous to skin, less carcinogenic and less toxic to nature. Most of them are water-soluble so they are also used as food colorant. With increasing use of chemical dyes, these days, there has also been remarkable increase in air, soil and water pollution. So there has been immediate need to produce eco-friendly dye from plant resources. This naturally obtained dyes are not only useful for pollution reduction but can also be used as edible products as food colouring agent.

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